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AThe vapor pressure data of polyaromatic hydrocarbons and the extrapolation of thedata using Antoines $\ln p$ function are presented in this Appendix. The temperaturedependence of the vapor pressure is given by the Clausius-Clapeyron equation: $\ln p(T) = \ln p_0 - \frac{H_s}{RT}$, where p_0 is the vapor pressure as $T \rightarrow \infty$, and H_s is the enthalpy of sublimationat temperature T . Due to the commonly observed increa...

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166 J. Chem. Eng. Data 1990, 35, 166-168Thermodynamic Properties of Aqueous Electrolyte Solutions. 1.Vapor Pressure of Aqueous Solutions of LiCl, LiBr, and LiI Kashinath R. Patil, Atri D. Tripathi, Gopal Pathak, and Sushilendra S. Katti National Chemical Laboratory, Pune 4 1 1008, IndiaThe vapor pressures of aqueous soiutions of lithium Table I. Vapor Pressures (P , kPa) of Aqueous Solutionshalide...

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